



## low impact development technical workshop series

### Water Quality Treatment in Bioretention

#### *Topics*

**Mechanisms**

**Performance**

**Special considerations**

# All primary pathways for removing pollutants from storm flows are active in bioretention



**Stormwater volume reduction.**

**Sedimentation.**

**Filtration.**

**Phytoremediation.**

**Thermal attenuation.**

**Adsorption.**

**Volatilization.**

**water quality treatment**

## Some characteristics of urban pollutants



**PAH's product of incomplete combustion and sealers. Coal tar emulsions may be 5-600x higher in PAH's concentrations than asphalt emulsion.**

**Annual loading of oil to Puget Sound ~22,580 metric tons (Exxon Valdez spilled ~33,500 metric tons).**

**Many pollutants associated with fines (particularly metals), many <0.45 microns (dissolved).**

**Ranges of metals from various studies:  
Zn (20-2000 µg/l) > (~344 metric tons/yr to PS)  
Cu ~Pb (5-200 µg/l) >  
Cd (<12 µg/l)  
(Davis et al 2001)**

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## Flow volume reductions in bioretention

	Completed	Infiltration	Sizing	Volume Reduction (%)
Siskiyou Green Street	Oct 2003	1.5 -2.0 in/hr	6%	*(1/04 – 12/05) 83%
Glencoe Rain Garden	Oct 2003	1.8 - 3.0 in/hr	6%	(1/04 – 12/05) 94%
Greensboro NC	2001	0.2 – 0.6 in/hr	5%	(2002) 78%
Sea Street	2001	variable		(2001 – present) 98%
110 <sup>th</sup> Cascade	2003			(10/04 – 06) 74%
Meadow on the Hylebos	2006	0.0 – 0.8 in/hr	15%	(10/07 – 5/08) 99.99%

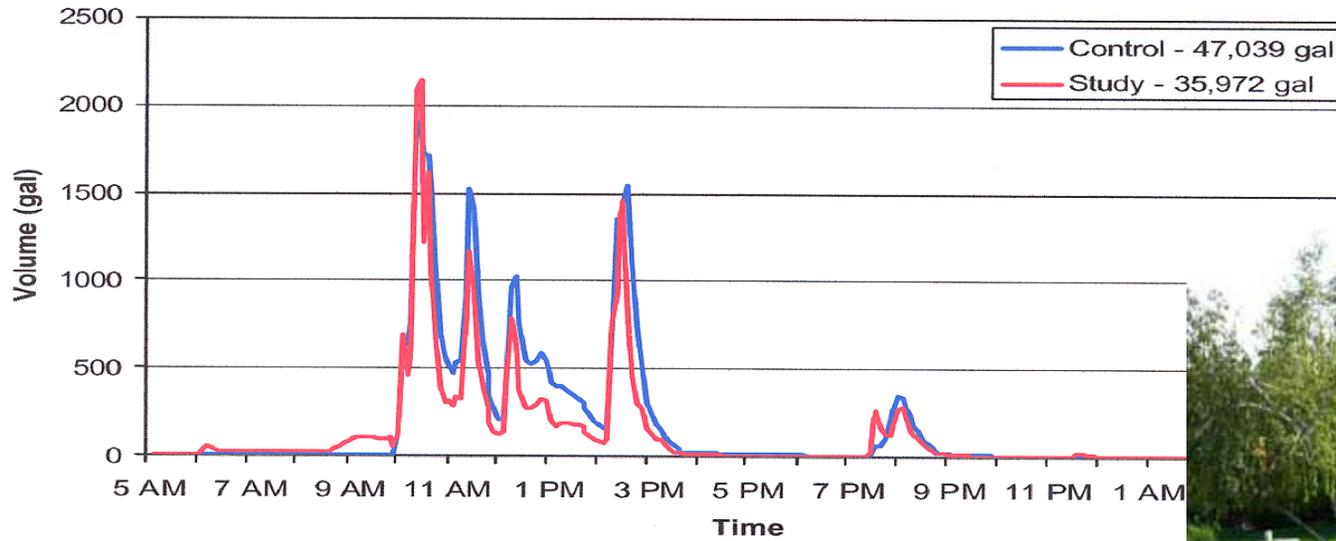


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### Pre-Construction Runoff Data

June 6, 2003

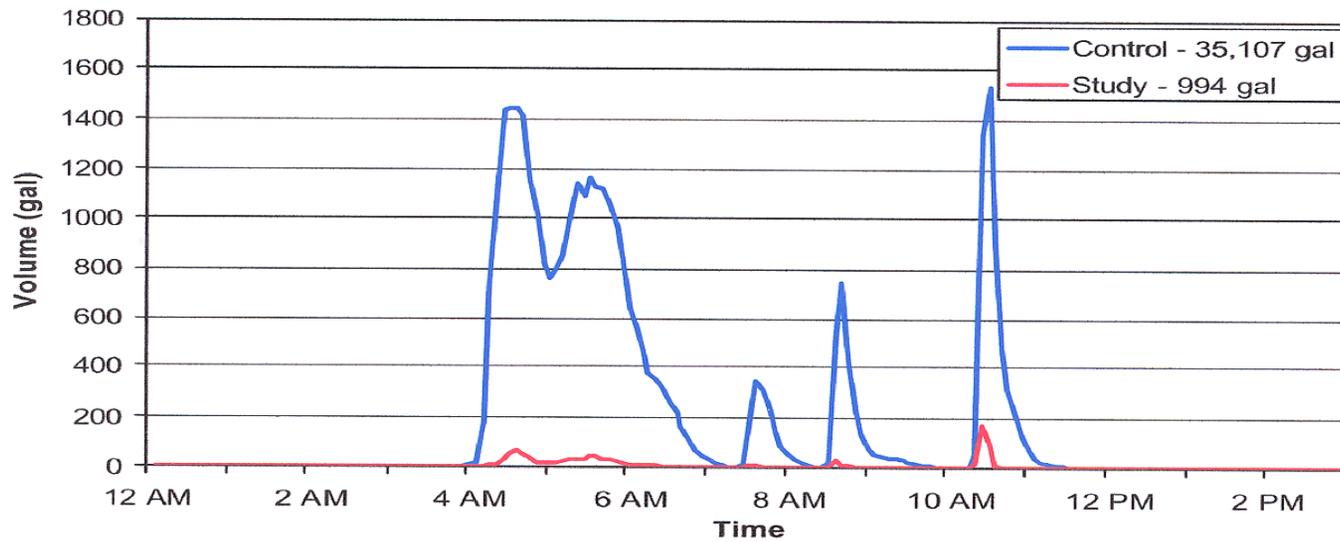
0.50" Rainfall



### Post-Construction Runoff Data

May 29, 2004

0.71" Rainfall



## Soil sampling in bioretention facilities

	e. Coli (mpn/g)	Cu (mg/kg)	Pb (mg/kg)	Hg (mg/kg)	Zn (mg/kg)
Siskiyou Green Street					
0-6"	280	34.4	56.8	0.103	170
6-12'	--	17.0	12.2	0.032	100
12-18"	--	17.6	10.9	0.054	96
SW 12 <sup>th</sup> & Montgomery					
0-6"	7	30.1	29.9	0.043	120
12-18"	--	22.2	18.9	0.082	92

MTCA  
 Pb: 250 mg/kg  
 Hg: 2 mg/kg



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## Percent removal of metals and TSS in bioretention and grass bioswales

	TSS (mg/L)	Cu ( $\mu\text{g/L}$ )	Pb ( $\mu\text{g/L}$ )	Zn ( $\mu\text{g/L}$ )
Davis et al 2001*		89% (u) 92% (l)	>98% (u) >98 (l)	>98% (u) >98 (l)
Davis et al 2003**		>99%	>99%	>99%
Greenbelt		97%	>95%	>95%
Largo		43%	70%	64%
Hunt et al 2006				
Greensboro	-180%	99%	81%	98%
Chapel Hill	--	--	--	--
Hsieh, Davis 2005	91%			
Multhanna et al 2007		63%	93%	87%
PNW Bioswales (Herrera 2006)	64%			47%
National Bioswales (Herrera from Barrett)	43%			53%

Event mean concentrations

\* Percent reduction at 18 cm (upper) and 61 cm (lower) depths (lab)

\*\* Percent mass removal (lab)

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## Percent removal of nutrients in bioretention and grass bioswales

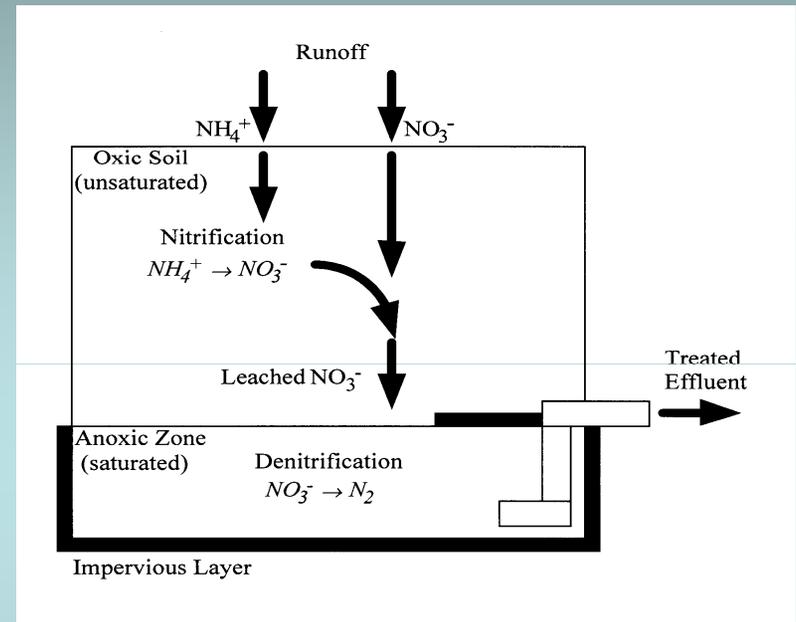
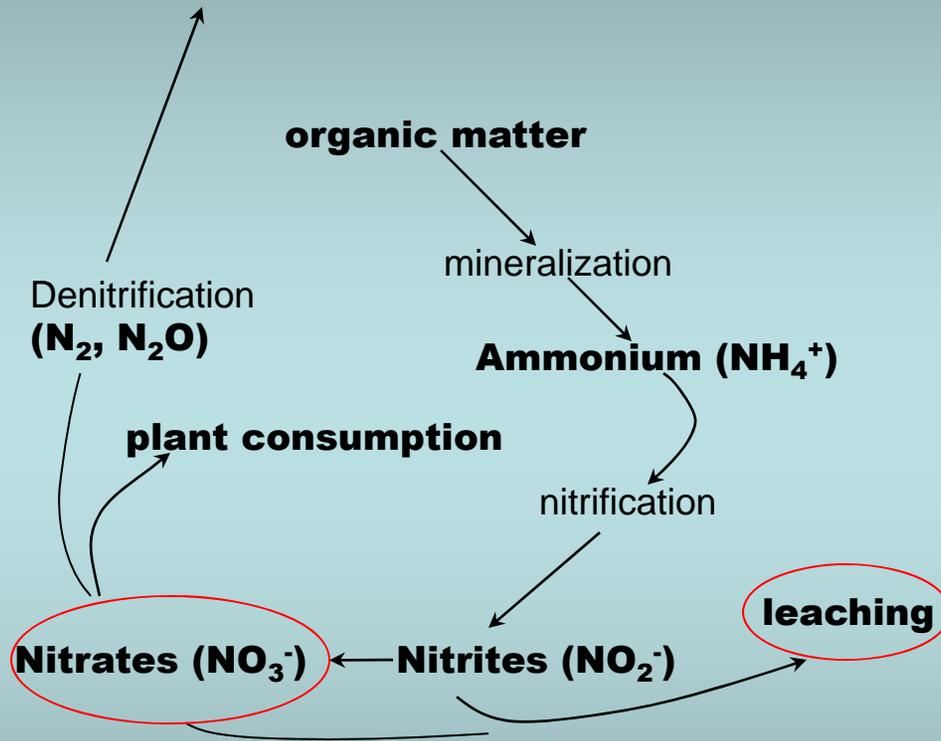
	TKN (mg/L)	NO <sub>3</sub> (mg/L)	TP (mg/L)	Hydrocarbons (µg/L)
Davis et al 2006*	38% (u) 68% (l)	-96% (u) 24% (l)	1% (u) 81% (l)	
Greenbelt	57%	16%	65%	
Largo	67%	15%	87%	
Mass removal	97%	97%	99%	
Hunt et al 2006				
Greensboro	-4.9%	75%	-240%	
Chapel Hill	45%	13%	65%	
Hsieh 2005				>97%
PNW Bioswales (Herrera 2006)			18%	-10%
Nat'l Bioswales**				-88%

Event mean concentrations

\* Percent reduction at 18 cm (upper) and 61 cm (lower) depths (lab)

\*\* Herrera from Barrett

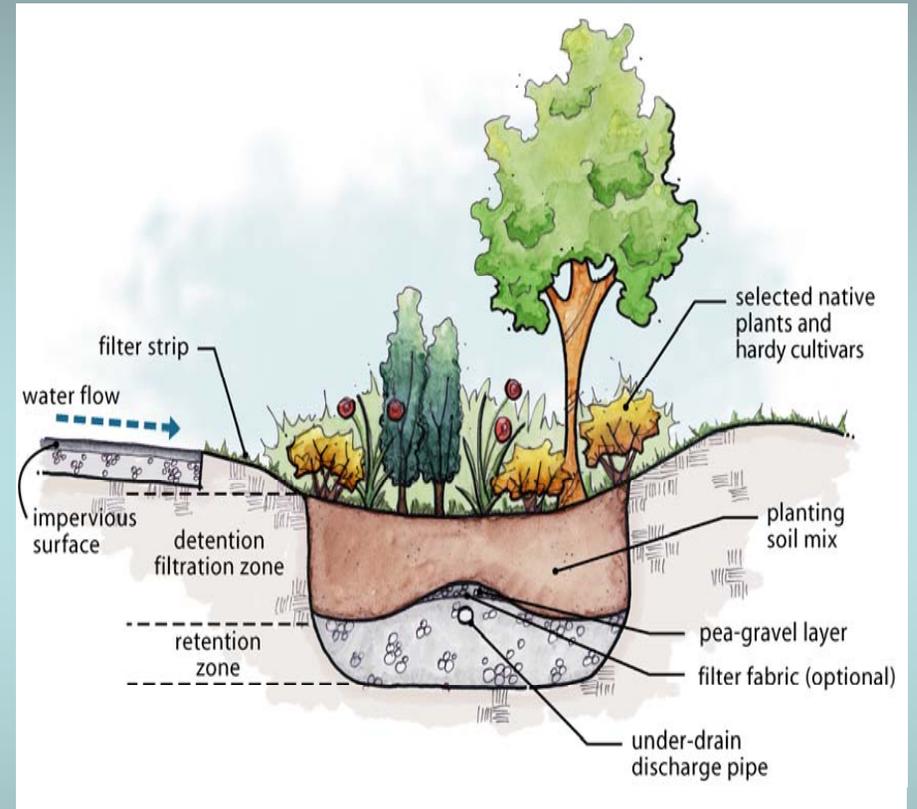
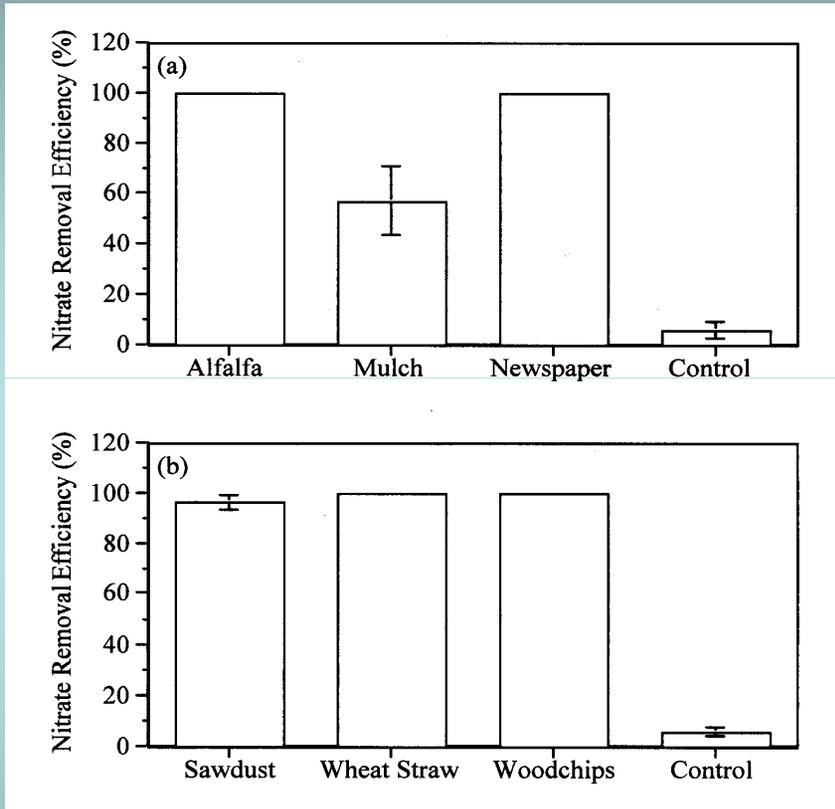
# Methods for managing nitrate



**$\text{NO}_3^-$  electron acceptor not  $\text{O}_2$  in anaerobic conditions**  
 $2\text{NO}_3^- + 10\text{e}^- + 12\text{H}^+ \rightarrow \text{N}_2 + 6\text{H}_2\text{O}$   
**Electron donor may be sugar, hydrocarbon (simple) or complex (mulch).**

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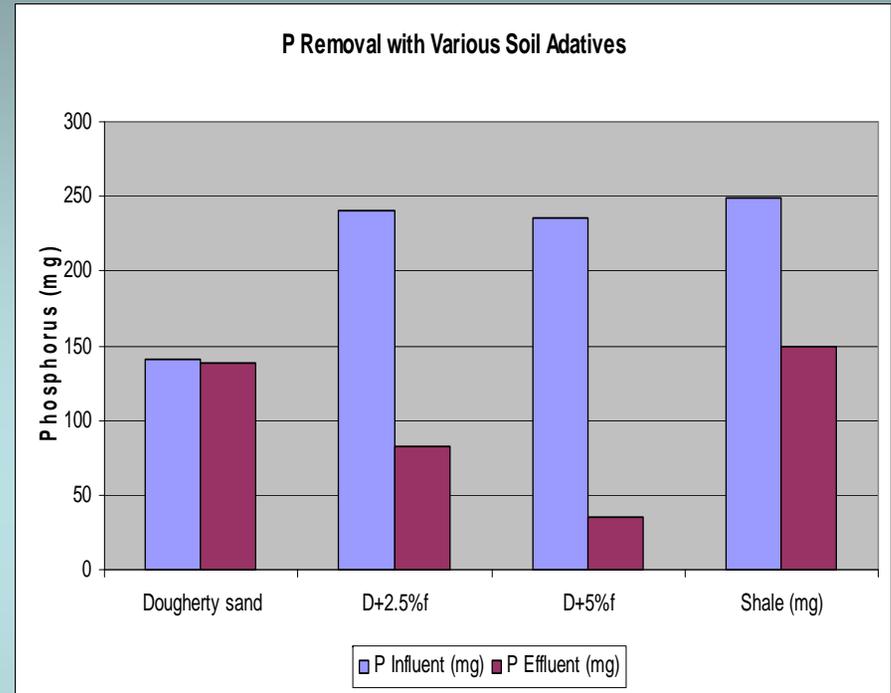
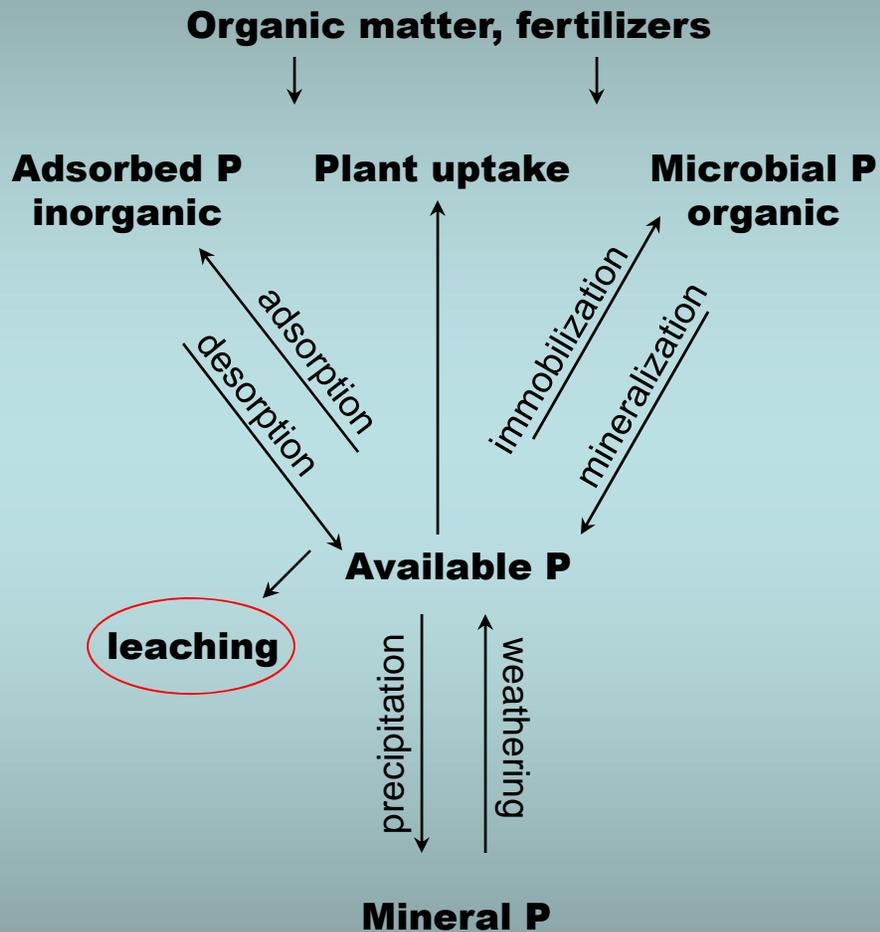
# Methods for managing nitrate



**Newspaper and woodchips good electron donor and carbon source for denitrification. (Kim Seagren, Davis, 2000)**

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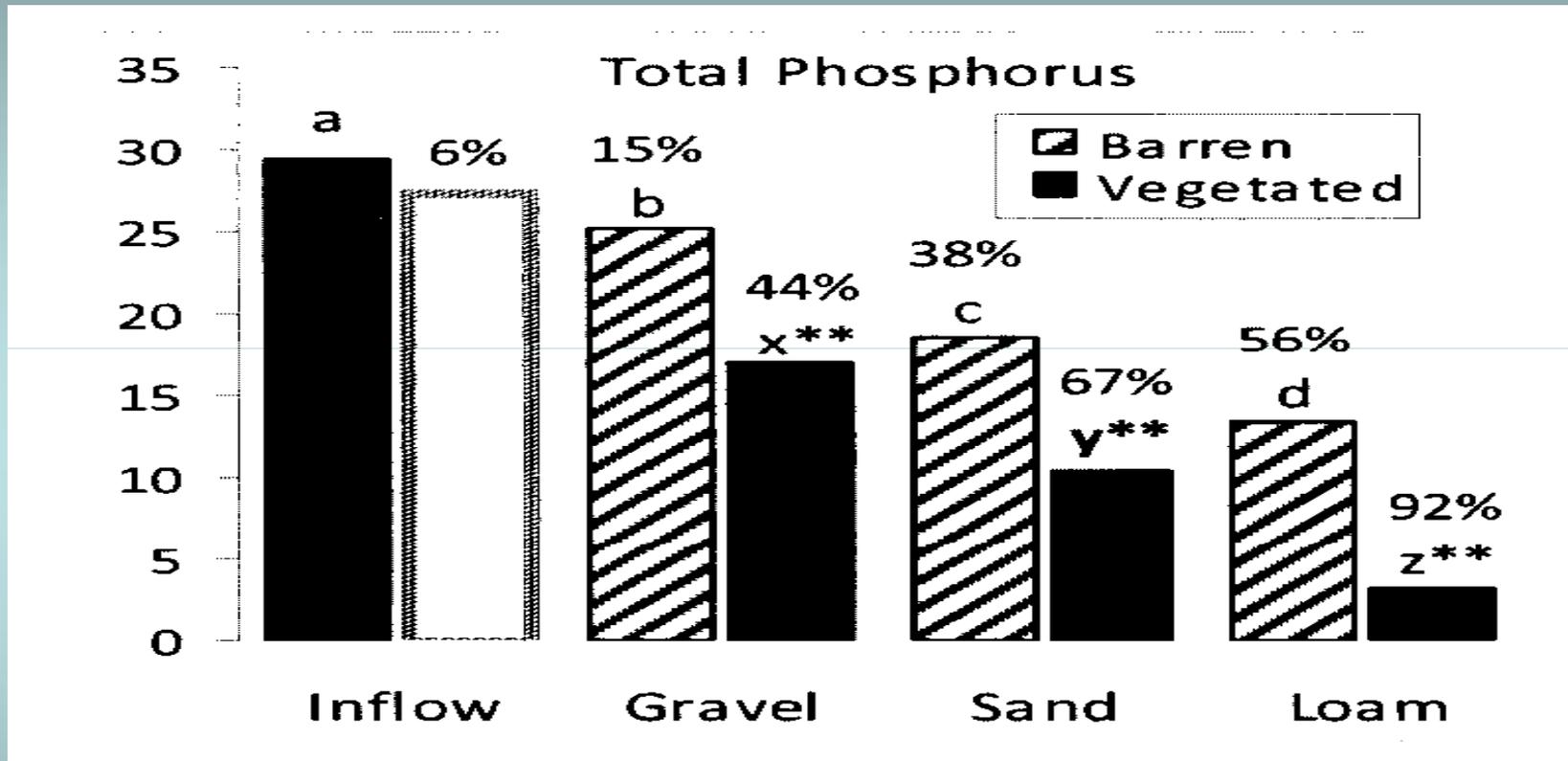
# Methods for retaining phosphate



**Fly ash significantly improves P retention, but significantly reduces K. (Zhang 2000)**

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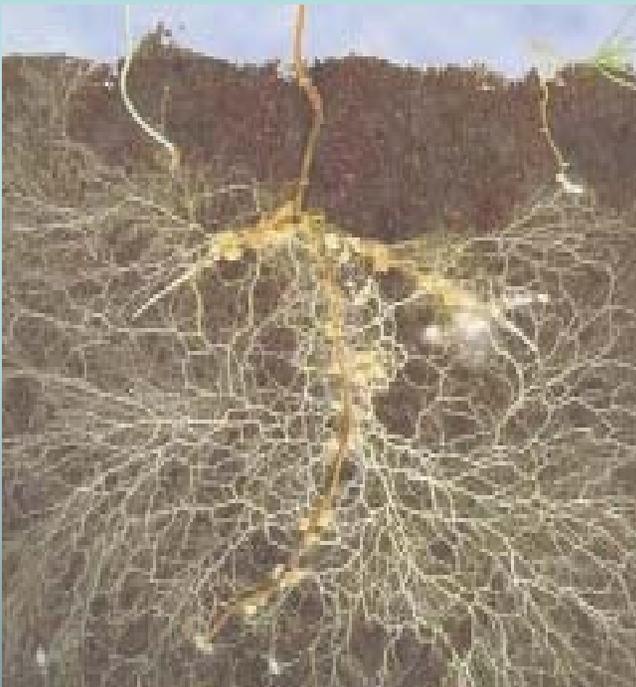
## Methods for retaining phosphate



**P-sorption increased significantly in vegetated vs non-vegetated plot. Increased  $O_2$  → oxidizes Fe and arbuscular mycorrhizal fungi possible mechanisms. (Lucas, Greenway 2007)**

## **Plants play a critical role in bioretention flow and water quality treatment performance**

- **Plant roots penetrate soil creating flow paths, exude saccharides and dead material that feed soil organisms and create soil aggregates.**



- **Treatment mechanisms:**
  - **Nutrient uptake.**
  - **Metal uptake.**
  - **Uptake, volatilization, transformation of organics.**
- **Plants influence water quality directly (e.g. uptake) and indirectly through physical and chemical changes to rhizosphere.**

## Summary and recommendations

- **Bioretention areas provide excellent metal, hydrocarbon and TSS removal.**
- **Metal, hydrocarbon and TSS removal primarily in upper few centimeters. Hydrocarbons transformed within a few days. Mulch layer most important for metal and hydrocarbon removal.**
- **Phosphorus and nitrogen removal is variable. Nitrate and phosphate export is possible. Removal likely improves with facility depth.**
- **Nitrate removal dependent on  $O_2$ . Use raised under-drain to create an anaerobic zone and improve  $NO_3$  for effluent release to marine water. Controlling HRT with under-drain orifice has not examined adequately yet.**

## Summary and recommendations

- **Phosphate removal primarily driven by sorption capacity (other factors include HRT, BSM depth, and possibly plants and microbial activity).**
- **More research needed for optimizing for phosphate and nitrate removal. WSU starts research program winter 2009.**
- **Discussion focused on percent removal and concentrations. When considering volume reduction in rain gardens, loads dramatically reduced for all constituents.**